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Eight new and eight known 2-(2-phenylethyl)chromone $(=2-(2-\text{phenylethvl})-4H-1-\text{benzopyran}-4$ one) derivatives, i.e., $1-8$ and $9-16$, respectively, together with the two known sesquiterpenoids 17 and 18 were isolated from a 70% MeOH extract of *Aquilaria malaccensis* (AM) agarwood chips. Their structures were determined based on extensive spectroscopic analysis and comparison with reported data.

Introduction. – Aquilaria malaccensis (Thymelaeaceae) is a species of a mediumsized evergreen tree found in the rain forests of Southeast Asia and New Guinea. This species, as other species of the *Aquilaria* genus, can produce a dark aromatic resin mainly in response to the infection by *Phaeoacremonium parasiticum*, a parasitic ascomycetous mould, which results in a resin-suffused heartwood well known as agarwood [1]. The agarwood has been used as incense as well as sedative, analgesic, and digestive medicine in traditional herb therapies [2]. Previous phytochemical research on this genus led to the isolation of many (phenylethyl)chromone derivatives $[2-10]$ and sesquiterpenoids [11 – 15]. Our ongoing investigation of A. malaccensis agarwood led to the isolation of eight new and ten known compounds, i.e., $1-8$ and $9-18$, respectively (Fig. 1). In this article, we present the isolation and structure elucidation of them.

Results and Discussion. – The 70% MeOH extract of A. malaccensis agarwood was partitioned with Et₂O, BuOH, and H₂O, as previously reported [15]. The BuOH fraction and $Et₂O$ fraction were then each subjected to repeated column chromatographic separation to afford eighteen compounds, including $1 - 8$ as new 2-(2phenylethyl)chromone $(=(2-\text{phenylethyl})-4H-1-\text{benzopyran}-4-\text{one})$ derivatives.

Compound 1 was obtained as brown amorphous powder, and its molecular formula $C_{19}H_{18}O_5$ was deduced from HR-ESI-MS (*m/z* 327.1232 ([*M* + H]⁺). The IR absorption at 1637 cm⁻¹ indicated the presence of a trisubstituted γ -pyrone (=4H-pyran-4-one) [3]. In the aromatic region of the ¹H-NMR spectrum (*Table 1*), two ss at δ (H) 7.52 and 6.93 (each 1 H) and two ds at $\delta(H)$ 7.09 and 6.80 (each $J = 8.6$ Hz, 2 H) were attributable to a tetrasubstituted and a para-disubstituted benzene ring, respectively. Besides that, a comprehensive consideration of the ¹H- and ¹³C-NMR data (*Table 1*) revealed the presence of one $C=O$, one olefinic CH, two CH₂, one OH, and two MeO

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Fig. 1. New and known compounds, i.e., $1 - 8$ and $9 - 18$, respectively, isolated from Aquilaria malaccensis agarwood

groups. The above findings suggested a (phenylethyl)chromone skeleton for 1, which was further confirmed by the ${}^{1}H,{}^{1}H$ -COSY and HMBC data (*Fig. 2*). The H-atom at $\delta({\rm H})$ 7.52 should be assigned to H–C(5) because of its HMBC cross-peak with C(4)=O. Three signals at δ (C) 145.1, 151.2, and 158.2 arising from O-bearing aromatic C-atoms were attributable to $C(6)$, $C(7)$, and $C(4')$ respectively, by related HMBC and comparison with a structurally similar known compound [4]. Two MeO groups were determined to be positioned at $C(6)$ and $C(4')$, based on the HMBC cross-peaks between the MeO H-atoms and corresponding C-atoms. Thus, the OH group should be linked to the last open position at $C(7)$. Consequently, the structure of 1 was elucidated as shown in Fig. 1.1)

Compound 2 was obtained as pale yellow amorphous powder, and its molecular formula was determined to be $C_{18}H_{18}O_6$ by HR-FAB-MS (m/z 331.1187 ($[M + H]^+$). Its

¹) Trivial atom numbering; for systematic names, see *Exper. Part.*

Position	1		$\mathbf{2}$	
	$\delta(H)$	$\delta(C)$	$\delta(H)$	$\delta(C)$
C(2)		167.9		169.2
$H-C(3)$	6.06(s)	109.4	6.10(s)	114.2
C(4)		177.7		180.8
$H - C(5)$	7.52(s)	104.3	3.76 $(d, J = 4.4)$	49.5
$C(6)$ or H–C(6)		145.1	3.83 $(d, J = 4.4)$	54.8
$C(7)$ or H–C(7)		151.2	4.10 $(d, J = 7.7)$	72.8
$H-C(8)$	6.93(s)	102.7	4.69 $(d, J = 7.7)$	69.0
C(9)		152.8		158.5
C(10)		117.0°		120.3
C(1')		131.8		131.1
$H-C(2')$	7.09 $(d, J=8.6)$	129.2	7.06 $(d, J = 8.5)$	129.1
$H-C(3')$	6.80 $(d, J=8.6)$	114.0	6.83 $(d, J = 8.5)$	114.2
C(4')		158.2		158.4
$H - C(5')$	6.80 $(d, J=8.6)$	114.0	6.83 $(d, J = 8.5)$	114.2
$H-C(6')$	7.09 $(d, J=8.6)$	129.2	7.06 $(d, J = 8.5)$	129.1
CH ₂ (7')	2.96 (dd, $J = 8.2, 7.2$)	32.2	2.91(m)	32.0
CH ₂ (8')	2.85 (dd, $J = 8.2, 7.2$)	36.3	2.82(m)	35.4
$MeO-C(6)$	3.97(s)	56.5		
$MeO-C(4')$	3.76(s)	55.3	3.77(s)	55.3

Table 1. ^{*IH*}- and ¹³C-NMR Data (CDCl₃, 500 MHz) of Compounds **1** and **2**¹). δ in ppm, *J* in Hz

a) The chemical shift was obtained from the HMBC coupling position (the peak of this C-atom could not be seen in the 13C-NMR spectrum due to the weakness of the signal).

Fig. 2. Key ¹H,¹H-COSY (\rightarrow) and HMBC (H \rightarrow C) features of **1, 2, 5**, and **8**

¹³C-NMR spectrum (*Table 1*) closely resembled that of oxidoagarochromone B (=(+) $rel-(1aR,1bR,2aR,6bR)$ -1a,1b,2a,6b-tetrahydro-4-[2-(4-methoxyphenyl)ethyl]-6H-bisoxirenol[f,h] [1]benzopyran-6-one) reported by Honda and co-workers [2], except for three strongly downfield-shifted signals at δ (C) 72.8, 69.0, and 54.8, probably due to the presence of a vicinal diol group instead of an epoxy group in this region. This speculation was also supported by the molecular mass of 2, which showed an increase of 18 mass units compared to that of oxidoagarochromone B. Careful examination of the ${}^{1}H, {}^{1}H$ -COSY (*Fig. 2*) and HSQC data (not shown) allowed the assignment of four consecutive O-bearing CH groups from $C(5)$ to $C(8)$, as shown in Table 1. The H-atom resonating at $\delta(H)$ 3.76, located at one end of this chain, was assigned to H–C(5) because of its correlation with the $C=O$ in the HMBC spectrum. When considering the relative configuration of 2, a half-chair conformation with minimum total energy was assumed for the cyclohexene ring (ring A). The large J value (7.7 Hz) of H–C(8) and H-C(7) indicated their *trans* axial relationship, whereas a very small coupling constant (almost zero) between H–C(7) and H–C(6) could be explained by the approximate 90° dihedral angle between them, which could be achieved by an equatorial orientation of H-C(6). This configuration was further confirmed by the NOESY correlation between $H-C(6)$ and $H-C(7)$. Since the O-atom took an axial position at $C(6)$, the epoxide could be formed only when this atom took the *cis*-equatorial position at $C(5)$, thus the H–C(5) must be axially oriented. Therefore, assuming a β -orientation of H–C(8), all of the other three H-atoms in ring A should be positioned on the α -face. Based on all these informations, the structure of 2 was determined as depicted in Fig. 1.

Compounds 3 and 4 possessed the same molecular skeleton as 2, including the relative configuration, as indicated by their ${}^{1}H$ - and ${}^{13}C$ -NMR spectra (*Table 2*). The differences among them appeared in the substitution pattern of benzene ring B. In the ¹H-NMR spectrum of 3, the *m*s at δ (H) 7.16 (2 H), 7.22 (1 H), and 7.30 (2 H) corresponded to an unsubstituted phenyl group, compared to a disubstituted one in 4 as demonstrated by a diagnostic peak combination at $\delta(H)$ 6.60 (dd, $J = 8.0$ and 1.6 Hz, $H-C(6')$), 6.74 (d, $J=1.6$ Hz, $H-C(2')$), and 6.75 (d, $J=8.0$ Hz, $H-C(5')$). The full

Position	3		4	
	$\delta(H)$	$\delta(C)$	$\delta(H)$	$\delta(C)$
C(2)		169.0		169.2
$H - C(3)$	6.12(s)	114.2	6.10(s)	114.2
C(4)		180.7		180.8
$H - C(5)$	3.76 $(d, J = 4.2)$	49.5	3.77 $(d, J = 4.2)$	49.5
$H-C(6)$	3.83 $(d, J = 4.2)$	54.7	3.83 $(d, J=4.2)$	54.8
$H-C(7)$	4.10 $(d, J = 7.7)$	72.9	4.10 (d, $J = 7.8$)	72.9
$H - C(8)$	4.69 $(d, J = 7.7)$	69.0	4.69 (d, $J = 7.8$)	69.0
C(9)		158.5		158.5
C(10)		120.3		120.3
C(1')		139.1		132.3
$H-C(2')$	7.16(m)	128.2	6.74 $(d, J = 1.6)$	114.3
$H-C(3')$ or $C(3')$	7.30 (m)	128.8		145.8
$H-C(4')$ or $C(4')$	7.22(m)	126.8		145.4
$H - C(5')$	7.30 (m)	128.8	6.75 $(d, J=8.0)$	110.8
$H-C(6')$	7.16 (m)	128.2	6.60 (dd, $J = 8.0, 1.6$)	119.6
CH ₂ (7')	2.98(m)	32.9	2.88(m)	32.2
CH ₂ (8')	2.86(m)	35.1	2.82(m)	35.2
$MeO-C(4')$			3.85(s)	56.0

Table 2. ¹H- and ¹³C-NMR Data (CDCl₃, 500 MHz) of Compounds **3** and 4^1). δ in ppm, *J* in Hz.

assignment of the H- and C-atom signals was accomplished on the basis of HMBC spectra and comparison with reported data of similar compounds [2].

Compound 5 was obtained as pale brown amorphous powder, and its molecular formula $C_{18}H_{19}ClO_6$ was deduced from the HR-ESI-MS (m/z 367.0931 ($[M+H]^+$, $C_{18}H_{20}ClO_6^+$). The existence of a Cl-atom could be confirmed by the *ca*. 3:1 peakintensity ratio of the ions $[M+H]^+ / [M+2+H]^+$. The diagnostic signals at $\delta(H)$ 6.82 $(d, J = 8.5 \text{ Hz}, 2 \text{ H})$ and 7.06 $(d, J = 8.5 \text{ Hz}, 2 \text{ H})$ in the ¹H-NMR spectrum (*Table 3*) indicated the presence of a 4-substituted phenyl group. Careful examination of the ¹ Hand 13C-NMR data led to the identification of one olefinic and three O-bearing CH groups, together with one MeO and two $CH₂$ groups, suggesting the existence of a highly oxygenated (phenylethyl)chromone skeleton [3]. This speculation was further confirmed by the analysis of the ${}^{1}H,{}^{1}H$ -COSY, HSQC, and HMBC data. Comprehensive consideration of the ${}^{1}H, {}^{1}H$ -COSY (*Fig. 2*) and HSQC data (not shown) allowed the assignment of four consecutive CH groups from $C(5)$ to $C(8)$, as shown in Table 3. The H-atom resonating at $\delta(H)$ 4.90, located at one end of this chain, was assigned to H–C(5) because its signal correlated with that of C(4) in the HMBC spectrum (*Fig.* 2). The linkage of $C(8)$ to a Cl-atom was revealed by the upfield shift of its signal in the ¹³C-NMR spectrum (*Table 3*), comparing to those of the O-bearing C-atoms. The MeO group was connected to $C(4')$, as indicated by a corresponding HMBC cross-peak (*Fig. 2*). Based on the above analysis, the planar structure of 5 could be established unambiguously. When considering the relative configuration of 5, a half-chair conformation with minimum total energy was assumed for the cyclohexene ring (ring

Position	5		6	
	$\delta(H)$	$\delta(C)$	$\delta(H)$	$\delta(C)^a$
C(2)		169.7		172.2
$H - C(3)$	6.08(s)	113.6	6.07(s)	115.1
C(4)		180.6		182.3
$H - C(5)$	4.90 $(d, J = 7.2)$	68.0	4.89 $(d, J = 7.2)$	67.3
$H-C(6)$	4.25 $(dd, J=7.2, 2.1)$	70.4	4.25 (dd, $J = 7.2$, 2.2)	74.4
$H - C(7)$	4.36 (dd, $J = 3.5, 2.1$)	73.2	4.36 (dd, $J = 3.5, 2.2$)	74.4
$H-C(8)$	4.71 $(d, J = 3.5)$	53.9	4.71 $(d, J = 3.5)$	58.9
C(9)		157.3		161.9
C(10)		121.2		123.7
C(1')		131.2		134.7
$H - C(2')$	7.06 $(d, J=8.5)$	129.2	6.74 $(d, J = 2.0)$	117.2
$H - C(3')$	6.82 $(d, J=8.5)$	114.2		148.5
C(4')		158.4		148.4
$H - C(5')$	6.82 $(d, J=8.5)$	114.2	6.74 $(d, J = 8.2)$	113.8
$H-C(6')$	7.06 $(d, J=8.5)$	129.2	6.60 (dd, $J = 8.2, 2.0$)	121.4
CH ₂ (7')	2.91(m)	32.0	2.88(m)	34.0
CH ₂ (8')	2.82(m)	35.6	2.81(m)	37.2
$MeO-C(4')$	3.77(s)	55.3	3.85(s)	57.3

Table 3. ^{*IH*}- and ¹³C-NMR Data (CDCl₃, 500 MHz) of Compounds **5** and 6^1). δ in ppm, *J* in Hz.

^a) The ¹³C-NMR spectrum of 6 was recorded in CD₃OD due to the poor solubility of this compound in CDCl₃

A). The large *J* value (7.2 Hz) of H-C(5) and H-C(6) indicated their *trans*-axial relationship, whereas a small coupling constant (2.1 Hz) between H–C(6) and H–C(7) could be explained by the axial-equatorial relationship of them. The absence of a NOESY interaction between H–C(6) and H–C(8) suggested the pseudoequatorial orientation of H–C(8). Therefore, assuming a β -configuration of H–C(8), H–C(5) should also be β -oriented, while H–C(7) and H–C(6) were positioned on the α -face. Consequently, the structure of 5 was determined as shown in Fig. 1.

Compound 6 was isolated as white amorphous powder. Its HR-ESI-MS exhibited a $[M+H]^+$ peak at m/z 383.0888 (C₁₈H₂₀ClO₇⁺) in accordance with the molecular formula $C_{18}H_{19}ClO_7$. The structure of 6 was very similar to that of 5, except for one more OH group substituting ring B. The NOESY correlations $H-C(2')/CH_2(7)$, H–C(6')/CH₂(7'), H–C(5')/MeO–C(4'), as well as the absence of a NOESY coupling H-C(2')/MeO-C(4') established that the OH group was attached to C(3'). The full assignment of the H- and C-atom signals, as shown in Table 3, was accomplished on the basis of 2D-NMR spectra and by comparison with 5.

An inseparable pair of compounds 7/8 was obtained as brown amorphous powder. The molecular formulas of 7 and 8 were determined to be $C_{17}H_{18}O_5$ (m/z 303.1220) $([M+H]^+, C_{17}H_{19}O_5^+)$ and $C_{18}H_{20}O_6$ (*m/z* 333.1326 ([*M*+H]⁺, C₁₈H₂₁O₆⁺), respectively, by HR-ESI-MS. Two sets of signals, arising from these two compounds, were observed in the ¹H-NMR spectrum (*Table 4*), with differences mainly in the aromatic region. The three ms resonating at $\delta(H)$ 7.27 (2 H), 7.19 (1 H), and 7.14 (2 H) were attributed to an unsubstituted phenyl group in 7, while two correlated ds at $\delta(H)$ 7.05 $(2 H)$ and 6.81 (2 H) could be ascribed to a *para*-substituted one in **8**. The skeletons of 7

Position	7		8	
	$\delta(H)$	$\delta(C)$	$\delta(H)$	$\delta(C)$
C(2)		168.8		168.9
$H-(3)$	6.10(s)	113.0	6.08(s)	113.0
C(4)		180.5		180.5
$H-(5)$	4.88 (br. s)	65.8	4.88 (br. s)	65.8
$H-(6)$	4.05 (br. s)	69.6	4.05 (br. s)	69.6
$H-(7)$	4.08 (br. s)	67.8	4.08 (br. s)	67.8
CH ₂ (8)	$2.77(m)$, 2.94 (m)	33.5	2.77(m), 2.94(m)	33.5
C(9)		162.3		162.3
C(10)		119.8		119.6
C(1')		139.4		131.4
$H-(2')$	7.14 (m)	128.2	7.05 $(d, J=8.5)$	129.1
$H-(3')$	7.27(m)	128.6	6.81 $(d, J=8.5)$	114.1
$H-(4')$ or $C(4')$	7.19(m)	126.6		158.3
$H-(5')$	7.27(m)	128.6	6.81 $(d, J=8.5)$	114.1
$H-(6')$	7.14 (m)	128.2	7.05 $(d, J=8.5)$	129.1
CH ₂ (7')	2.92(m)	32.9	2.86(m)	32.0
CH ₂ (8')	2.79(m)	35.1	2.76(m)	35.4
$MeO-C(4')$			3.75(s)	55.2

Table 4. ¹H- and ¹³C-NMR Data (CDCl₃, 500 MHz) of Compounds **7** and $\mathbf{8}^1$). δ in ppm, *J* in Hz.

and 8 could easily be established as (phenylethyl)chromones on the basis of their ¹ Hand 13 C-NMR spectra. In both 7 and 8, a chain consisting of three consecutive Obearing CH groups and one CH₂ group was identified with the assistance of the ${}^{1}H,{}^{1}H$ -COSY and HSQC experiments, as shown in Fig. 2. The signal at $\delta(H)$ 4.88 in the ¹H-NMR spectrum was assigned to $H-C(5)$ because of its correlation with the $C(4)$ signal in the HMBC spectrum $(Fig, 2)$. Based on all these information, the planar structures of 7 and 8 were characterized as shown in Fig. 1. Assuming a half-chair conformation for the ring A of **7** (or **8**) (*Fig. 3*), the orientation of H–C(7) should be equatorial since no large coupling constant was observed between this H-atom and either H-atom at $C(8)$. The H-C(6) was also equatorially oriented, as revealed by the absence of a NOESY interaction $H-C(6)/CH_2(8)$. For stability reasons, it was most indicated to propose an axial orientation for $H-C(5)$ to avoid the presence of three axial OH groups in ring A . The equatorial OH $-C(5)$ was able to form a H-bond with the C=O group at $C(4)$ [6] [7], which further stabilized this conformation. Therefore, the relative configurations of 7 and 8 were unambiguously established (*Fig. 1*).

Fig. 3. Conformation of the cyclohexene moiety (ring A) of compound 7 (or 8)

The absolute configurations of compounds $2-8$ were not determined due to the scarcity of the samples.

The known compounds isolated were identified as 6-methoxy-2-(2-phenylethyl) chromone (9) [4] [10], 6,7-dimethoxy-2-(2-phenylethyl)chromone (10) [4] [10], 6,7dimethoxy-2-[2-(4-methoxyphenyl)ethyl]chromone (11) [8], 5,6 : 7,8-diepoxy-5,6,7,8 tetrahydro-2-(2-phenylethyl)chromone (12) [2], 5,6 : 7,8-diepoxy-5,6,7,8-tetrahydro-2- [2-(4-methoxyphenyl)ethyl]chromone (13) [2], 5,6 : 7,8-diepoxy-5,6,7,8-tetrahydro-2- [2-(3-hydroxy-4-methoxyphenyl)ethyl]chromone (14) [2], 2-(2-phenylethyl)chromone (15) [5] [9], 6-hydroxy-2-(2-phenylethyl)chromone (16) [4], neopetasane (17) [14], and baimuxinal (18) [13] by comparing their spectroscopic data with the reported ones. Among them, $12 - 14$ were obtained for the first time from the A. malaccensis species.

Experimental Part

General. Column chromatography (CC): silica gel 60 (70-230 mesh (No. 107734); 230-400 mesh (No. 109385); Merck) and SephadexTM LH-20 (GE Healthcare, Sweden). Anal. TLC: silica gel 60 F_{254} (SiO₂; Merck). Semi-prep. HPLC: Hitachi instrument with a Gemini RP C₁₈ column (250 × 10 mm, 5 um). Optical rotations: Jasco-P-1020 polarimeter (Japan). UV Spectra: Hitachi-U-3010 spectrophotometer (Japan); λ_{\max} (log ε) in nm. IR Spectra: *Jasco-FT/IR-4200* spectrometer (Japan); $\tilde{\nu}$ in cm⁻¹. NMR Spectra: Bruker-Avance (500 MHz) FT-NMR spectrometer (Germany); δ in ppm rel. to Me₄Si as internal standard, J in Hz. MS: Bruker-micrOTOF-Q-II mass spectrometer (Germany); in m/z.

Plant Material. The agarwood chips of A. malaccensis were purchased from Industrial Plantation Co., Vientiane, Laos, in January 2010. The voucher specimen was deposited with the Herbarium of the Natural Product Research Institute, Seoul National University.

Extraction and Isolation. Air-dried A. malaccensis agarwood chips $(450 g)$ were crushed and exhaustively extracted by refluxing with 70% MeOH. The extract was concentrated to give a residue of 68.2 g, which was successively partitioned with Et₂O, BuOH, and H₂O [15]. The BuOH fraction (31.0 g) was subjected to CC (SiO₂, AcOEt/MeOH 50 : $1 \rightarrow 10$: 1): *BFrs.* 1 – 5. *BFr.* 2 was further separated by CC (SiO₂, hexane/AcOEt 2:1 \rightarrow 1:2): BFrs. 2.1 - 2.9. BFr. 2.1 was applied to CC (SiO₂, hexane/CHCl₃ $10 : 1 \rightarrow 3 : 1$): 9 (17 mg). *BFr. 2.2* was subjected to CC (SiO₂, CHCl₃/AcOEt 20:1): **10** (15 mg). *BFr. 2.3* was applied to CC (SiO₂, hexane/CHCl₃ 10 : 1 \rightarrow 1 : 1): **11** (9 mg). Compounds **1** (11 mg), **12** (28 mg), **13** (17 mg), and a mixture $2/3$ were obtained from BFr. 2.5 after CC (SiO₂, CHCl₃/AcOEt 20:1 \rightarrow 3:1). Separation of 2 and 3 was finally achieved by CC (SiO₂, CHCl₃): 2 (14 mg) and 3 (33 mg). BFr. 2.6 was subjected to CC (SiO₂, CHCl₃): **14** (25 mg). *BFr.* 2.7 was applied to CC (SiO₂, CHCl₃/AcOEt 20:1 \rightarrow 5:1): 4 (22 mg). BFr. 2.8 was applied to CC (SiO₂, CHCl₃/AcOEt 5:1 \rightarrow 3:1, CHCl₃/MeOH 40:1 \rightarrow 20 : 1): BFrs. 2.8.1 – 2.8.5. BFr. 2.8.3 was subjected to CC (SiO₂, CHCl₃/MeOH 100 : 1 \rightarrow 60 : 1): 5 (11 mg). BFr. 2.8.4 was applied to CC (SiO₂, CHCl₃/MeOH 80:1 \rightarrow 35:1): 6 (5 mg). BFr. 3 was separated by CC (SiO₂, CHCl₃/MeOH 40 : 1 \rightarrow 10 : 1): *BFrs.* 3.1 – 3.5. A mixture 7/8 (21 mg) was obtained from BFr. 3.3 by CC (SiO₂, CHCl₃/MeOH 80:1 \rightarrow 10:1). Although different solvent combinations and several kinds of columns (including C_{18} semi-prep. HPLC columns) were tried, separation of 7/8 was not successful due to the structural similarity of them. The Et_2O fraction (21.5 g) was also subjected to CC (SiO₂, hexane/AcOEt 40:1 \rightarrow 1:1): *EFrs. 1 – 14. EFr. 1* was purified by CC (*Sephadex LH-20*, MeOH): 17 (36 mg). Compounds 15 (42 mg) and 18 (12 mg) were obtained from *EFr. 4* after separation by CC (SiO₂, CHCl₃). Compound **16** (86 mg) was isolated from *EFr. 14* by recrystalization in MeOH. All the isolated compounds were further purified by CC (Sephadex LH-20) or semi-prep. reversed-phase HPLC.

7-Hydroxy-6-methoxy-2-[2-(4-methoxyphenyl)ethyl]-4H-1-benzopyran-4-one (1): Brown amorphous powder. UV (MeOH): 317 (3.65), 278 (3.71), 228 (4.15), 205 (4.18). IR (film): 2929, 1736, 1637, 1513, 1475, 1383, 1279, 1247, 1217, 1179, 1116, 1081, 1034. ¹H- and ¹³C-NMR: Table 1. HR-ESI-MS (pos.): 327.1232 ([$M + H$]⁺, C₁₉H₁₉O₅⁺; calc. 327.1227).

rel-(1aR,2R,3R,7bS)-1a,2,3,7b-Tetrahydro-2,3-dihydroxy-5-[2-(4-methoxyphenyl)ethyl]-7H-oxireno[f] [1]benzopyran-7-one (2): Pale yellow amorphous powder. [α] $_{15}^{25}$ = +18.0 (c = 0.178, MeOH). UV (MeOH): 253 (4.00), 223 (4.18), 203 (4.23). IR (film): 3398, 1658, 1600, 1513, 1458, 1247, 1184, 1060. ¹ Hand ¹³C-NMR: *Table 1*. HR-FAB-MS (pos.): 331.1187 ($[M+H]^+$, $C_{18}H_{19}O_6^+$; calc. 331.1182).

rel-(1aR,2R,3R,7bS)-1a,2,3,7b-Tetrahydro-2,3-dihydroxy-5-(2-phenylethyl)-7H-oxireno[f] [1]benzopyran-7-one (3). White amorphous powder. $\lbrack a \rbrack_0^2 = -38.9$ (c = 0.141, MeOH). UV (MeOH): 253 (4.03), 205 (4.24). IR (film): 3265, 2917, 1651, 1573, 1456, 1187, 1010. ¹H- and ¹³C-NMR*: Table 2*. HR-ESI-MS (pos.): 301.1066 ([$M + H$]⁺, C₁₇H₁₇O₅⁺; calc. 301.1071).

rel-(1aR,2R,3R,7bS)-1a,2,3,7b-Tetrahydro-2,3-dihydroxy-5-[2-(3-hydroxy-4-methoxyphenyl)ethyl]- 7H-oxireno[f] [1] benzopyran-7-one (4): Brown amorphous powder. $\lbrack \alpha \rbrack_0^{25} = +60.9$ (c = 0.183, MeOH). UV (MeOH): 204 (4.47). IR (film): 3388, 2931, 1658, 1604, 1513, 1442, 1246, 1186, 1131, 1032. ¹ H- and ¹³C-NMR: *Table 2*. HR-ESI-MS (pos.): 347.1110 ([$M + H$]⁺, C₁₈H₁₉O₇⁺; calc. 347.1125).

rel-(5R,6S,7S,8R)-8-Chloro-5,6,7,8-tetrahydro-5,6,7-trihydroxy-2-[2-(4-methoxyphenyl)ethyl]-4H-1 *benzopyran-4-one* (5): Pale brown amorphous powder. $\left[a\right]_D^{25} = -2.6$ ($c = 0.138$, MeOH). UV (MeOH): 253 (4.00), 203 (4.25). IR (film): 3399, 2917, 2849, 1734, 1657, 1610, 1513, 1438, 1247, 1180, 1099, 1033. ¹ Hand ¹³C-NMR: *Table 3*. HR-ESI-MS (pos.): 367.0931 ($[M + H]^+$, $C_{18}H_{20}ClO_6^+$; calc. 367.0943).

rel-(5R,6S,7S,8R)-8-Chloro-5,6,7,8-tetrahydro-5,6,7-trihydroxy-2-[2-(3-hydroxy-4-methoxyphenyl)ethyl]-4H-1-benzopyran-4-one (6): White amorphous powder. $\lbrack a\rbrack_5^5 = -3.9$ (c = 0.172, MeOH). UV (MeOH): 253 (3.91), 203 (4.37). IR (film): 3354, 2918, 2849, 1734, 1657, 1590, 1513, 1437, 1276, 1241, 1130, 1099, 1024. ¹H- and ¹³C-NMR*: Table 3*. HR-ESI-MS (pos.): 383.0888 ([$M+H$]⁺, C₁₈H₂₀ClO \ddagger ; calc. 383.0892).

rel-(5R,6S,7R)-5,6,7,8-Tetrahydro-5,6,7-trihydroxy-2-(2-phenylethyl)-4H-1-benzopyran-4-one (7) and rel-(5R,6S,7R)-5,6,7,8-Tetrahydro-5,6,7-trihydroxy-2-[2-(4-methoxyphenyl)ethyl]-4H-1-benzopyran-4-one (8): Brown amorphous powder (mixture). $\lbrack a \rbrack_0^{25} = -10.6$ (c = 0.221, MeOH). UV (MeOH): 250, 203. IR (film): 3397, 2918, 2848, 1733, 1658, 1601, 1513, 1439, 1246, 1179, 1032. ¹H and ¹³C-NMR*: Table 4*. HR-ESI-MS (pos.): 303.1220 ($[M+H]^+$, C₁₇H₁₉O₂[†]; calc. 303.1227) for **7** and 333.1326 ($[M+H]^+$, $C_{18}H_{21}O_6^+$; calc. 333.1333) for 8.

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